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Kelly R. Lab results: 1 doz grains of rice = 0.3 g (Use this fact as a conversion factor.)? grains of rice = 1.94 g

Avogadro's Number - the mole = the number 6.022×10^{23} Molar Mass the mass of one mole of any pure substance molar mass of an element (g/mol) equals the atomic mass of the element in grams molar mass of a compound equals the sum of the molar masses of the elements making up the compound . example problems: Find the molar masses of ...

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rice = g (Use this fact as a conversion

factor.) ? grains of rice = 1.94 g

Avogadro's Number - the mole = the

number 6.023×10^{23} Molar Mass the

mass of one mole of any pure substance

molar mass of an element (g/mol)

equals the atomic mass of the element

in grams molar mass of a compound

equals the sum of the molar mass of the

element making up the compound

example problems ...

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Introduction to the Mole and Molar Mass:

The terms mole and molar mass are introduced and used in calculations.

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How many formula units are in 32.6 grams of potassium oxide?

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A Study of Matter: Physical and Chemical Properties and ...

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