

Chemistry Chapter 14

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Chemistry Chapter 14

Chemistry - Chapter 14 1. An acid is used in car batteries. 2. The active ingredient of vinegar is an acid. 3. Acids are used for cleaning metals.

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1 . One example for NH_3 as a conjugate acid: $\text{NH}_2^- + \text{H}^+ + \text{NH}_3$; $\text{NH}_2^- + \text{H}^+ + \text{NH}_3$; as a conjugate base: $\text{NH}_4^+ + (\text{aq}) + \text{OH}^- (\text{aq}) \rightarrow \text{NH}_3 (\text{aq})$. (a) $\text{NaHSeO}_3 < \text{NaHSO}_3 < \text{NaHSO}_4$; in polyoxy acids, the more electronegative central element—S, in this case—forms the stronger acid. The larger number of oxygen atoms on the central atom (giving it a higher oxidation state) also creates ...

Answer Key Chapter 14 - Chemistry 2e | OpenStax

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Chapter 14 - Acids and Bases | CourseNotes

NCERT INTEXT QUESTIONS. 14.1 Glucose or sucrose are soluble in water but cyclohexane and benzene (simple six membered ring compounds) are insoluble in water Explain. Ans. The solubility of a solute in a given solvent follows the rule ' Like dissolves like'. Glucose contains five and sucrose contains eight -OH groups.

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CHAPTER 14 REVIEW Acids and Bases SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Name the following compounds as acids: sulfuric acid a. H_2SO_4 sulfurous acid b. H_2SO_3 hydrosulfuric acid c. H_2S perchloric acid d. HClO_4 hydrocyanic acid e. hydrogen cyanide 2. H_2S Which (if any) of the acids mentioned in item 1 are binary acids? 3.

14 Acids and Bases - David Brearley High School

14.114.114.1 Carbohydrates Chemistry 412 in milk is known as lactose. Carbohydrates are also called saccharides (Greek: sakcharon means sugar). Carbohydrates are classified on the basis of their behaviour on hydrolysis. They have been broadly divided into following three groups.

Biomolecules

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Chemistry-Chapter 14 > Across. The energy change that accompanies a chemical or physical change. The entropy of the universe can never decrease. It must always stay the same or increase. The study of the relationships and conversations between different forms of energy. The energy required to break a mole of a given type of bond ...

Chemistry-Chapter 14 - Crossword Puzzle

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